



Anhydrous HCL Gas Generator



Assembly Over Glass Lined Reactor



Pilot Scrubber



We form a combination of Chemical Engineers in the field of Process engineering and its application in the glass equipments/process packages.

### DESIGN AND SUPPLY OF VARIOUS

**UNITS** as per client's requirement are as under:

1. Absorption systems for gases such as HCl, Cl<sub>2</sub>, SO<sub>2</sub>, HBr, NH<sub>3</sub>, Br<sub>2</sub>, NO<sub>x</sub> etc.
2. Anhydrous HCl Gas Generation Unit by different routes viz.
  - Sulphuric Acid Route      - Boiling Route
  - Calcium Chloride Route    - Cyclic Route
3. HBr Gas Generator (By Boiling Route)
4. Sulphuric Acid Dilution Units.
5. Hypochlorite Manufacturing Units
6. MCA Condensation Assembly
7. Distillation - Conversion of Batch Process to Continuous Process.
8. Solvent Recovery
9. Iodine Recovery
10. Raw DCB Plant.

### KNOW-HOW WITH PLANT

for the following products :

1. Bromine recovery from industrial waste / salt bittern.
2. Nitric Acid Concentration
3. Sodium Hypochlorite

### TROUBLE SHOOTING

in above units as well as client's existing plant.

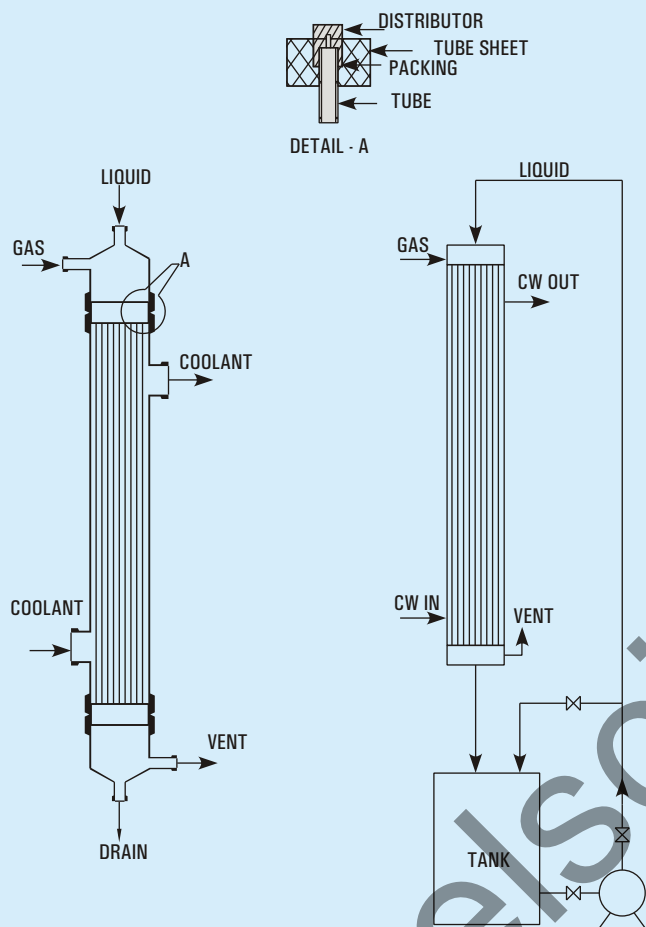
## FALLING FILM ABSORBER

Efficient gas absorption depends on the following :

1. Intimate contact.
2. Efficient Heat Transfer.

This is achieved in a Falling Film Absorber which is essentially a shell & tube heat exchanger in which both gas to be absorbed and absorbing liquid flow concurrently downward with extraction of heat by circulation of coolant in the shell. The absorbing liquid is circulated through a tank till desired concentration is achieved. The liquid flows at such a rate that the tubes do not flow full of the liquid but instead, descends by gravity along the inner walls of the tubes as a thin film. Obviously, this produces a much greater linear velocity for a given rate flow than could be obtained if the tube flowed full.

The equipment works as a number of water cooled wetted-wall columns in parallel and each tube is provided with distribution system on top to effect uniform distribution of both liquid and gas and also formation of a thin liquid film on the inner surface of the tube.



## SALIENT FEATURES

1. The heat of absorption is continuously removed. This ensures better absorption and product concentration as compared with conventional packed tower.
2. Low residence time and operating temperature ideally suited to heat sensitive materials.
3. Borosilicate glass and PTFE contact parts ensure corrosion/contamination free operation.
4. Both standard and custom built units are available.
5. Capable of operating from zero to maximum gas flow rate.
6. Ease of installation due to light weight.
7. Trouble free and consistent performance with minimal attention.
8. Wide application e.g. HCl, HBr, NH<sub>3</sub>, SO<sub>2</sub>, H<sub>2</sub>S, Br<sub>2</sub> etc.
9. Less cost.
10. Negligible pressure drop compared to conventional columns.
11. Compact design Sleek and slender.
12. Both heat and mass transfer operations are incorporated in a single equipment.
13. Very high heat transfer coefficient as the liquid falls instead of flowing.
14. Scaling of process fluid is minimal due to high velocity and ease of cleaning by simple acid circulation.
15. Hot conditions are eliminated at all stages namely pipe, tanks and pumps etc.

## LIMITATIONS

1. Not recommended for gases containing high proportion of inert (insoluble).
2. Not applicable if the gases are not highly soluble.

## SPECIFICATIONS

Sr. No.	Nominal Size (mm)	Absorber Area (m <sup>2</sup> )	No. of Tubes/ Tube OD (mm)	Max. Gas Absorption Rate (Pure HCl) * (kg/hr)	Max. Acid Prod. Rate (As 30 % HCl) (kg/hr) *	Height (m)
1.	80	1.00	4/ 20	30	100	4400
2.	100	1.76	7/ 20	60	200	4500
3.	150	4.80	19/ 20	150	500	4600
4.	225	7.80	31/ 20	250	833	4920
5.	300	15.30	61/ 20	500	1667	5050
6.	400	36.00	143/ 20	1175	3917	5300
7.	450	47.00	187/ 20	1500	5000	5700
8.	600	84.00	333/ 20	2700	9000	5800



# TECHNICAL PACKAGES

## SULPHURIC ACID CONCENTRATION SYSTEM

Commercial sulfuric acid is a cheap commodity and in its dilute and impure form does not have good market potential.

Disposal by neutralization with lime is associated with, problems of solid (calcium sulfate) handling and also adds to the cost.

To overcome these problems GOEL offers know-how, design engineering services of sulfuric acid concentration system in which the dilute sulfuric acid generated is reconcentrated to desired level for reuse. That is to say a twofold benefit of eliminating the disposal problem and minimizing fresh commercial acid requirement.

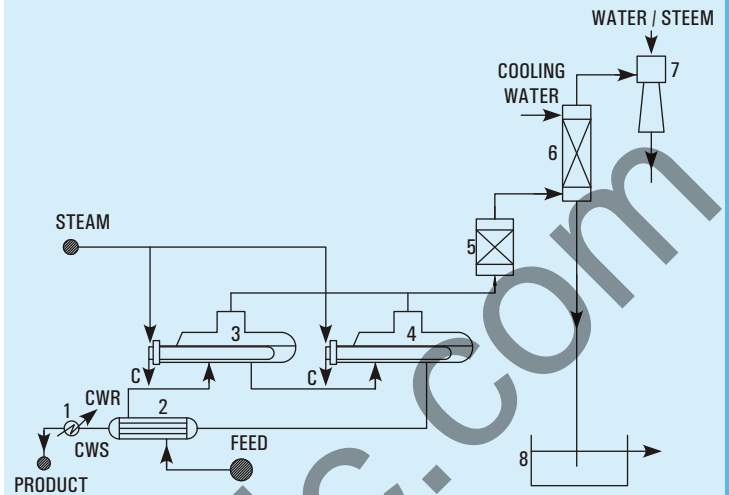
### PROCESS DESCRIPTION

The process is extremely simple & involves concentration of dilute sulfuric acid by evaporation using steam/thermic fluid as heating media under vacuum. The dilute feed is preheated & fed to a series of evaporators in stages to achieve the concentration level. The vapors from the evaporators are condensed and drained out through barometric legs and non-condensables are removed by an ejector. The final product is cooled and drawn in the storage tank.

### REQUIREMENT OF UTILITIES

The requirement of utilities viz. steam/thermic fluid, cooling water and product vary widely depending on feed rate, feed concentration and product concentration. For example for a plant having 50TPD feed containing 50% H<sub>2</sub>SO<sub>4</sub> the consumption of utilities for achieving 70% concentration are given below.

1. Steam @ 6 bar(g) pressure	800 Kg/hr
2. Cooling water @ 30 °C	70 Cu M/hr
3. Power	15 KW



TWO STAGE SULFURIC ACID CONCENTRATION SYSTEM

### KEY

- 1 - PRODUCT COOLER
- 2 - FEED PREHEATER
- 3 - 1ST, STAGE EVAPORATOR
- 4 - 2ND, STAGE EVAPORATOR
- 5 - MIST ELIMINATOR
- 6 - DIRECT COOLER
- 7 - EJECTION
- 8 - SEAL POT
- C - STEAM CONDENSATE
- CWS - COOLING WATER SUPPLY
- CWR - COOLING WATER RETURN



CWS - COOLING WATER SUPPLY  
 CWR - COOLING WATER RETURN  
 CHW - CHILLED WATER  
 R - ROTAMETERS  
 BATTERY LIMITS  
 GOEL CLIENT

## TECHNICAL PACKAGES

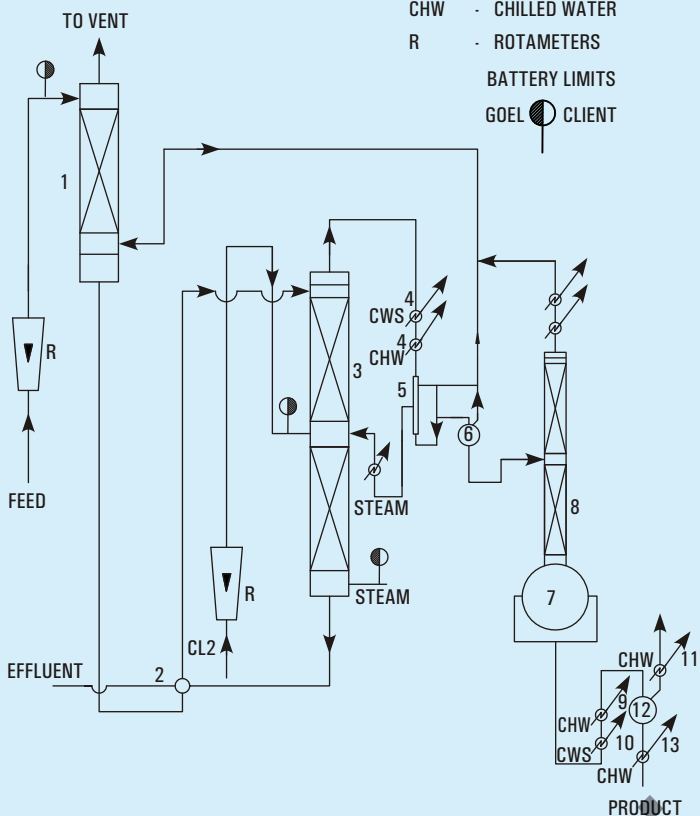
### BROMINE RECOVERY SYSTEM

Bromine is available in the sea bittren, as well as Industrial waste e.g. Aq. HBr / Aq. NaBr / Aq. KBr. The Bromine concentration in the feedstock varies from 2 gpl to 300 gpl from industry to industry.

Goel Offer suitable bromine recovery plant for the various feedstock based on his 20 years expertise in this field. Goel suggest cold process for bromine concentration below 3 gpl and Hot process above 3 gpl.

The package considered is schematically shown in Drawing enclosed herewith.

The process consists of simultaneous chlorination & steam blowing. The feed stock acidic in nature is preheated to near its boiling in feed pre heater and then fed to the main column where steam and chlorine are blown simultaneously. The liberated bromine as set free by chlorine are steam distilled. The liberated bromine together with steam and some excess chlorine is condensed in the condenser. The condensate is taken to a gravity separator where bromine and bromine water are separated. While bromine is taken in the purification column the aq. layer is recycled into the main column. Crude bromine is purified under reflux and pure bromine is collected in the receiver. All uncondensed vapour pass through the tail scrubber to recover the last traces of bromine.



SR.	DESCRIPTION
1.	TAIL SCRUBBER
2.	FEED PREHEATER
3.	Br <sub>2</sub> STRIPPING COLUMN
4.	Br <sub>2</sub> CONDENSERS
5.	PHASE SEPERATOR
6.	CRUDE Br <sub>2</sub> RECEIVING VESSEL

7.	REBOILER
8.	PURIFICATION COLUMN
9.	PRODUCT COOLER
10.	PRODUCT COOLER
11.	VENT CONDENSER
12.	PRODUCT RECEIVER VESSEL
13.	PRODUCT COOLER





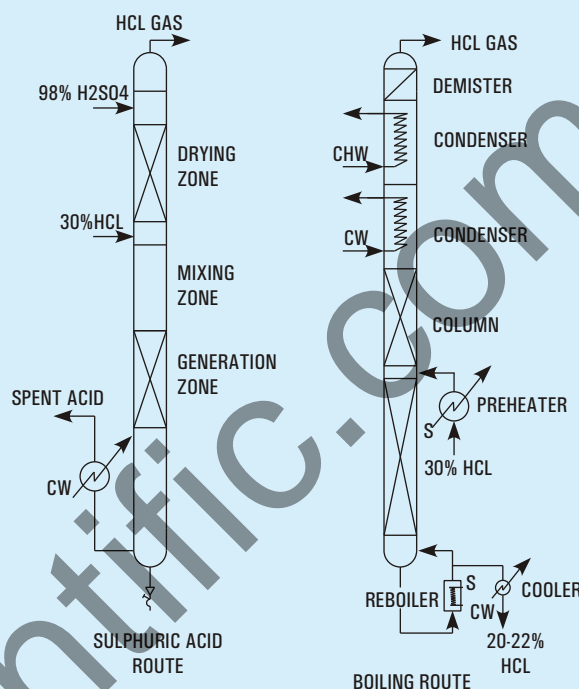
## ANHYDROUS HCL GAS GENERATOR

Commercial Hydrochloric Acid is available in the market as 30% aqueous solution. But for certain applications e.g. bulk drug and pharmaceuticals, HCl is required in anhydrous state for critical reactions where moisture cannot be tolerated. Such users generate anhydrous HCl from commercial grade for their captive consumption.

### METHOD

Several methods have been adopted by industries. But generation by Sulphuric Acid Route and Boiling Route are commonly practiced.

We offer Calcium Chloride Route also.



Route	Sulphuric Acid Route	Boiling Route
Working Principle	Hydrochloric acid is highly soluble in water but the solubility diminishes in presence of H <sub>2</sub> SO <sub>4</sub> and at 70 to 75% H <sub>2</sub> SO <sub>4</sub> concentration its solubility is negligible. Thus by adding (98%) commercial Sulphuric acid to commercial hydrochloric acid (30%) in proper ratio the entire HCl can be liberated in gaseous form leaving 75% H <sub>2</sub> SO <sub>4</sub> as spent acid.	Aqueous hydrochloric acid forms a maximum boiling point azeotrope at 110°C containing 20.24% HCl at atmospheric pressure. Thus by distilling commercial hydrochloric acid (30%) pure HCl gas can be generated and spent acid will contain over 20.24% HCl.
Process Outline	Metered quantities of commercial sulphuric acid hydrochloric acids are fed to the unit where they mix in the Mixing Zone. The gas generated forms a froth and enters the Generation Zone where while traveling through a bed gas is released which travels upwards through the Drying Zone. Here the gas comes in intimate contact with downward flow of 98% H <sub>2</sub> SO <sub>4</sub> . The dry gas leaving the unit passes through a rotameter. The spent liquor containing 70-75% H <sub>2</sub> SO <sub>4</sub> passes through the Cooling Zone before being discharged.	Metered quantity of commercial hydrochloric acid is preheated in a preheater by steam and fed to a fractionating column with steam as heating media in the reboiler. The vapours leaving the column are condensed with coolant as cooling water and chilled brine in stages. The relatively dry gas passes through a mist eliminator and then through a rotameter. The spent acid containing 22% HCl is cooled through a cooler and then discharged.
Salient Features	<ul style="list-style-type: none"> <li>- Operational reliability the unit can be started/ stopped in seconds.</li> <li>- Available in wide range of capacities from 5 to 200 kg/hr of dry HCl.</li> <li>- Except cooling water no other utility e.g. steam chilled water etc. required.</li> <li>- Anhydrous gas.</li> <li>- Capable of operating from 25 to 120%.</li> <li>- Ease of installation.</li> <li>- Negligible pressure drop.</li> <li>- High efficiency 99%.</li> </ul>	<ul style="list-style-type: none"> <li>- Operational reliability.</li> <li>- Available in wide range capacities from 5 kg/hr to 200 kg/hr of dry HCl.</li> <li>- Except commercial hydrochloric acid, no other raw-material is required.</li> <li>- Anhydrous gas.</li> <li>- Capable of operating from 25-100%.</li> <li>- Ease of installation.</li> <li>- Negligible pressure drop.</li> </ul>
Indicative Raw-material & Utilities for 20 kg/hr HCl	30% HCl - 70 kg/hr 98% H <sub>2</sub> SO <sub>4</sub> - 170 kg/hr Cooling Water - 2 m <sup>3</sup> /hr	30% HCl - 200 Kg/hr Saturated Steam - 50 kg/hr Cooling Water - 3.5 m <sup>3</sup> /hr Chilled Brine - 4 m <sup>3</sup> /hr

## TECHNICAL PACKAGES

### ANHYDROUS HCl GAS GENERATOR - CALCIUM CHLORIDE ROUTE

#### Working Principle:

Hydrochloric acid and water form a maximum boiling point azeotrope at 110°C corresponding to a concentration of 20.24% (w/w) HCl. By adding concentrated CaCl<sub>2</sub> solution to commercial hydrochloric acid the azeotrope point is eliminated and the entire HCl becomes available for liberation by distillation. Anhydrous HCl gas generation through Calcium Chloride Route is the most environmental friendly technique.

#### Process Description:

The above principle is achieved in practice by feeding metered quantities of commercial HCl and 50% CaCl<sub>2</sub> solution to a stripping column with a steam heated re-boiler at bottom. The effluent from bottom of the column is a dilute acidic calcium chloride solution which is concentrated to 50% in an evaporator and re-used. The vapor leaving is condensed stage wise with cooling water and chilled brine as coolant. The relatively dry gas passes through a mist eliminator and then through a rotameter to the point of consumption.

#### Raw material utility requirements:

The indicative requirements for 20 Kg/hr HCl gas generator are given below.

- |   |   |     |
|---|---|-----|
| 1. 30-32% HCl, (Kg/hr)                          | : | 66  |
| 2. Cooling water at 30 °C (M <sup>3</sup> /hr)  | : | 4   |
| 3. Chilled brine at -10 °C (M <sup>3</sup> /hr) | : | 3   |
| 4. Steam at 6 Kg/cm <sup>2</sup> (g)            | : | 150 |

#### LEGEND

R - REBOILER

C - COLUMN

E - EVAPORATOR

PH - PRIMARY CONDENSER

#### LEGEND

SH - SECONDARY CONDENSER

H - CONDENSER

ME - MIST ELIMINATOR

P - PUMP

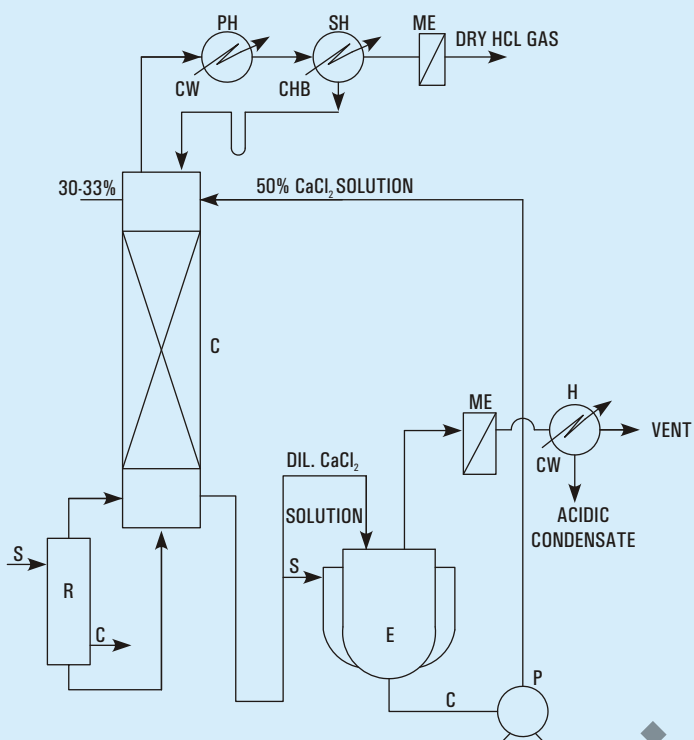
#### LEGEND

CW - COOLING WATER

CHB - CHILLED DRAINE

S - STEAM

C - CONDENSATE



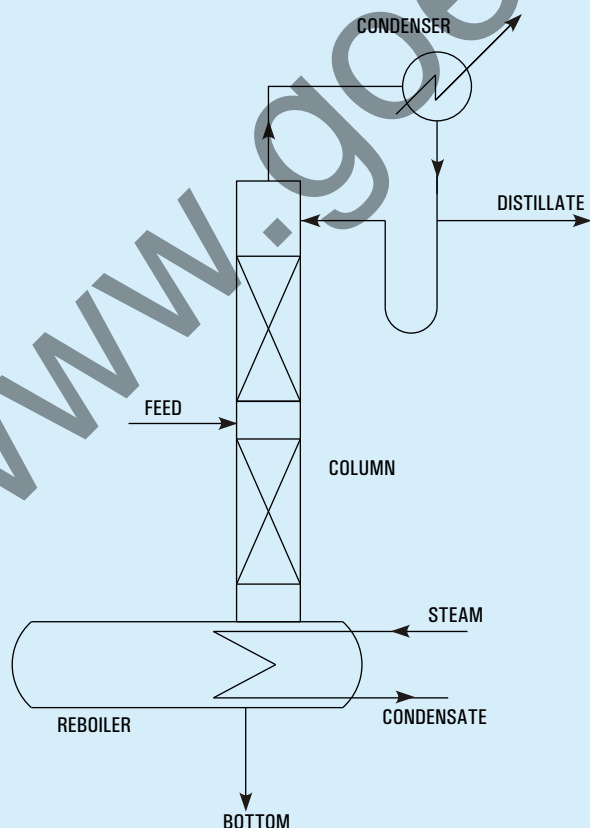
### CONTINUOUS DISTILLATION SYSTEM

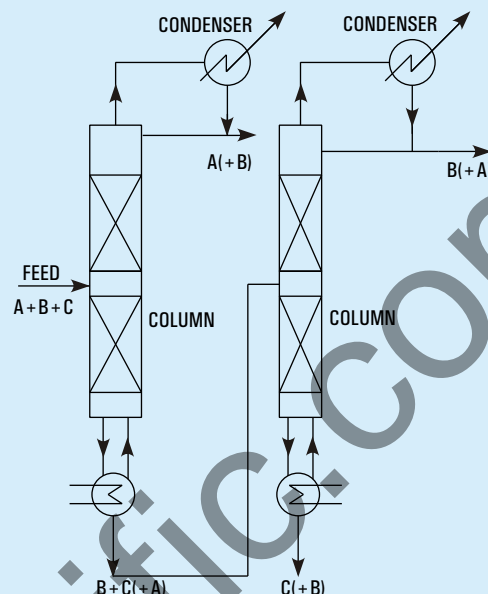
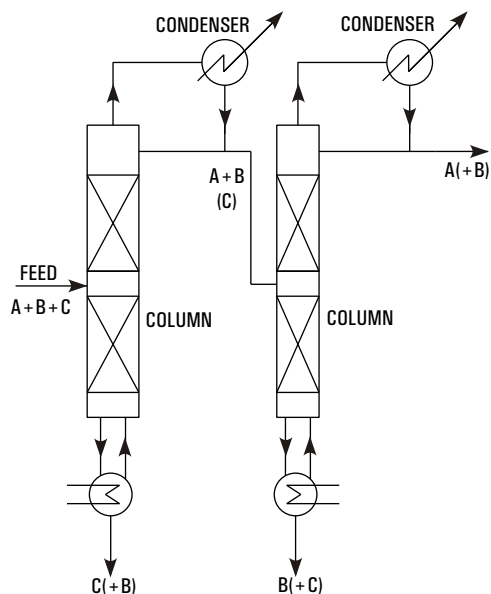


The limitations of batch distillations are naturally circumvented in continuous distillation as shown in fig.2 which is a typical fractionating unit comprising of ratification & stripping sections. Here feed is continuously fed to the column with withdrawal of top & bottom products. The process takes care on its own by simply maintaining the flow rates of feed & utilities.

However when more than two products are desired as in case of multicomponent systems additional columns are required as each column is capable of giving two products only. That is to say, for multicomponent system only one product is obtain in relatively pure form from each column. The other product containing the remaining components is fed to a subsequent column where again one product is obtained in relatively pure form. The addition of columns continue till the system becomes binary & both components are separated in the final column.

An important principle to be emphasized is that a total n-1 fractionators are required for complete separation of system of n components. Which of the two products in a column is to be obtained in relatively pure form depends on relative volatility of each component in the feed stock. For example consider a ternary solution consisting of a components A, B & C whose relative volatilities are in that order (A most volatile). In order to obtain three substances in substantially pure form either of the schemes shown in fig-3 may be used. Which of the two schemes would be used depends on the relative difficulties of separation in each method and the choice calls for finer considerations of principles of distillation. However scheme (b) is usually preferred since it requires one vaporization of substance A.





Component given in bracket are in small quantities.

## SOLVENT RECOVERY

Solvents are universally used in wide variety of industries, their use by no means being limited to the chemical industry. The choice of solvents such as xylene, acetone, butyl acetate, methanol etc. depends on the type of application and economical considerations.

In many cases, the solvent - after use - is contaminated and not in a reusable condition. Purchase of fresh solvent and disposal costs of the contaminated solvent can prove expensive operations. Therefore, it makes sound economic sense to investigate the recovery of contaminated solvents for re-use.

GOEL design and supply solvent recovery plants which are capable of recovering solvents from a few kg/day to many tons/day.

### Applications:

1. GOEL Solvent Recovery Units are carefully designed to cater the specific requirement for each duty for wide range of solvents.
2. Continuous Distillation Units have distinct advantages, such as
  - 2.1. Ease of Operation due to steady state working.
  - 2.2. Economical design as each equipment viz. column, condenser reboiler is designed for uniform and steady-load.
  - 2.3. Uniform product quality,
  - 2.4. Uniform & low consumption of utilities.
  - 2.5. Higher productivity as down time for start-up, emptying etc. is eliminated.

The outline flow-sheet of the plan is shown in Figure. Regulated quantity of feed is fed to a distillation column from an overhead vessel. The overhead vessel is provided with a reboiler with chilled water circulation to arrest loss of acetone vapor. The reboiler at the bottom of column in this case was a steam heated oil-bath. The vapors from top of the column pass through primary and secondary condensers with cooling water and chilled water circulation. The condensate is partially refluxed and balance continuously drawn out and collected in receiver provided with a vent condenser. The residue from the reboiler is continuously drained out.

